

## Stereoselective Synthesis of $\alpha,\alpha'$ -Bridged bis( $\alpha$ -Alanine) Derivatives.

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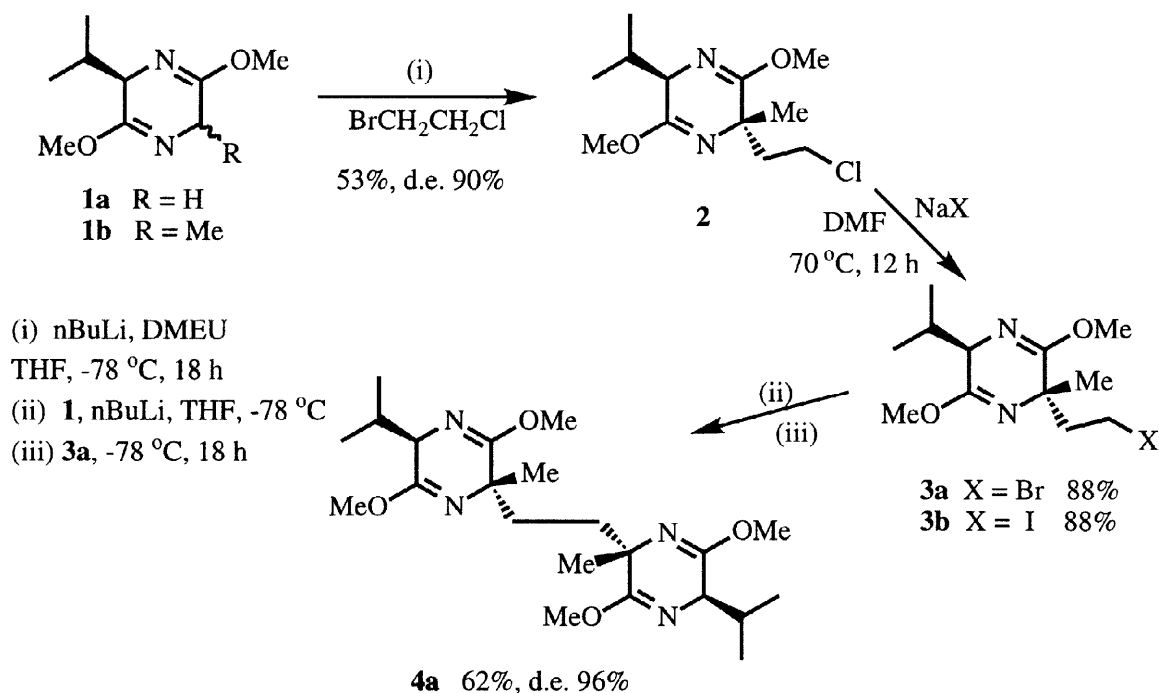
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**Abstract:** Stereoselective syntheses of  $\alpha,\alpha'$ -dimethylated C<sub>2</sub>–C<sub>4</sub> alkylene-bridged bis(glycine) methyl esters are described. The products were further converted into *N*-Fmoc-protected bis(amino acids).  
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The preparation of rigidified  $\alpha$ -amino acids to effect conformational constraints in peptides has assumed an important role in drug design and development.<sup>1</sup> Isosteric structures may be envisaged to replace cystine in peptoid bioorganics when cystine exerts mainly a structural skeletal function. We have recently reported syntheses of several bridged bis(glycines) in which the  $-\text{CH}_2\text{SSCH}_2-$  bridge between the two glycine units in cystine has been replaced by substituted even-numbered carbon bridges with special emphasis on C<sub>4</sub>-rigidified bridges as cystine substitutes with reduced conformational freedom.<sup>2</sup> C<sub>3</sub>-bridges, especially halogenated bridges and  $\alpha$ -hydroxymethyl derivatives, have received considerable attention.<sup>3</sup>



Scheme 1

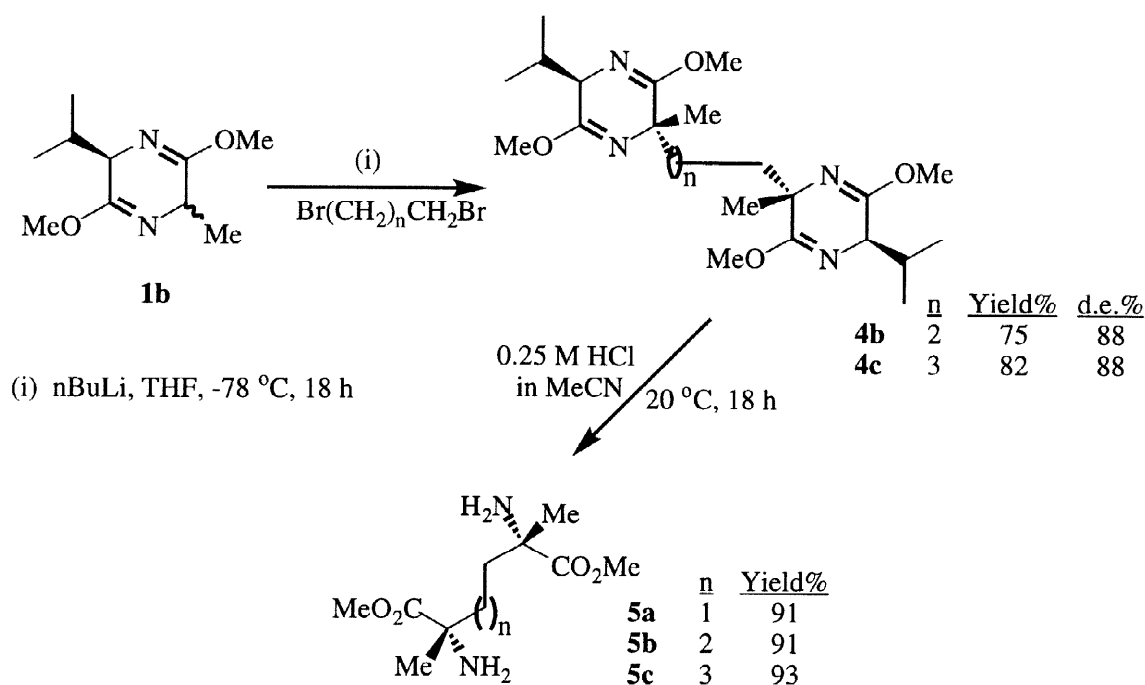
In an alternative approach to affect conformational preferences, the  $\alpha$ -hydrogen in  $\alpha$ -amino acids can be replaced by a carbosubstituent as exemplified by simple  $\alpha$ -methyl and higher  $\alpha$ -alkyl amino acid derivatives.<sup>4</sup> We have further increased the rigidity of  $\alpha$ -aminoid acid structures by incorporation of the  $\alpha$ -amino acid carbon into cyclic structures.<sup>5</sup> In this report we describe work on  $\alpha,\alpha'$ -dimethyl derivatives of bridged bis(glycines).

The chiral auxiliary to be used as pro-alanine was the 5-methylated derivative of (2*R*)-2,5-dihydro-3,6-dimethoxy-2-isopropylpyrazine (**1b**, Scheme 1). The latter is available by methylation of the lithiated bislactim ether **1a** or by *O*-methylation of the cyclic dipeptide prepared from (*R*)-valine and racemic alanine.<sup>6</sup> The stereochemistry at C-5 in substrate **1b** is of little importance since this stereochemistry is lost on metalation before the second C-5 alkylation. Stereoselectivity in the second alkylation step is trans with respect to the isopropyl group. In monoalkylations diastereomeric excess (d.e.) in the first alkylation is variable. In most dialkylations, however, the diastereoselectivity is often high with **1a** as the original substrate.<sup>5,6</sup> Stereochemistry at C-5 is controlled either by the stereochemistry of the isopropyl group at C-2, *ie.* the configuration of valine used for the bislactim ether, or by the order of the alkylation reactions using the same chiral auxiliary. In this work our efforts have been directed towards producing bis(alanines) with (*S*)-configuration at the stereogenic centers using **1b** as the substrate.

Alkylations were effected by initial lithiation by *n*-butyllithium at -78 °C followed by addition of the alkylating agent, an  $\alpha,\omega$ -dibromoalkane. For the C2-bridge, however, this approach may cause problems since vicinal dihalides are also agents used for halogenation at anionic carbon.<sup>7</sup> A trial with 1,2-dibromoethane for alkylation confirmed extensive halogenation and elimination reactions. This problem was largely overcome when the second halogen atom was chlorine, the reagent being either 1-bromo-2-chloroethane or 1-chloro-2-iodoethane. 1-Bromo-2-chloroethane was the better reagent. When the alkylation of lithiated **1b** was run at -78 °C in THF in the presence of 1,3-dimethyl-2-imidazolidinone (DMEU) with excess 1-bromo-2-chloroethane to yield the chloroethyl product **2**.

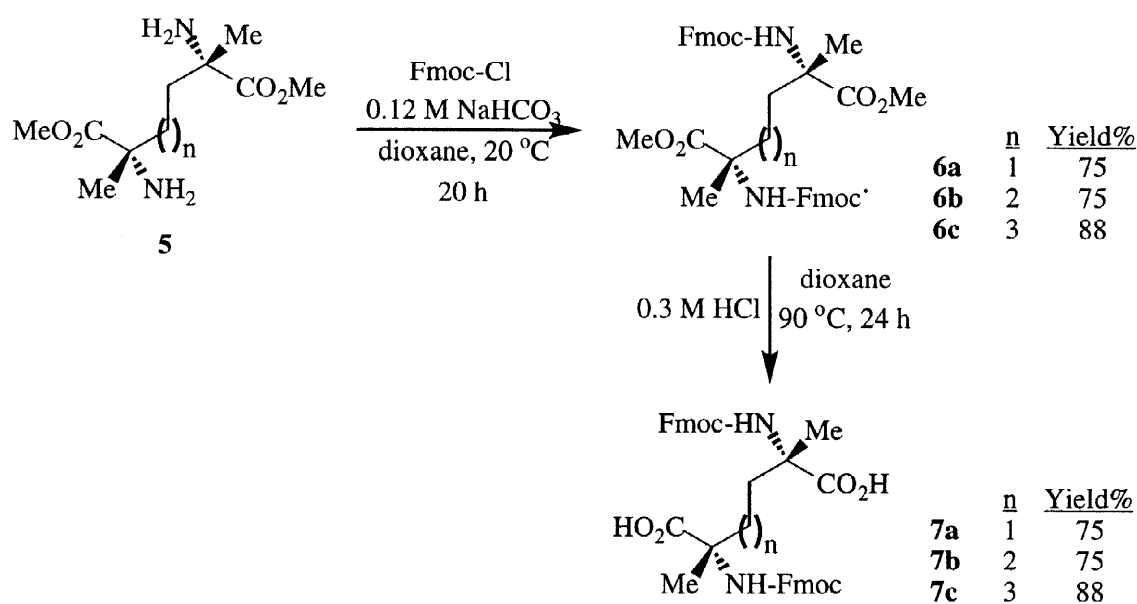
Prior to the second alkylation for the preparation of **4a**, halogen exchange was effected by warming the chloride **2** with either sodium bromide or iodide in DMF. Coupling of either the bromide **3a** or the iodide **3b** yielded the bridged structure **4a** which was isolated in 60% (d.e.96%) after flash chromatography which removes any diastereomeric product coming from the wrong stereoisomer of the alkylating agent **3a**. Coupling reactions with the other  $\alpha,\omega$ -dibromalkanes proceeded in high yields and were characterized by high diastereoselectivity. It is noteworthy that the high stereoselectivity results after two separate alkylation steps where in principle two different alkylating agents are involved in the formation of the product **4**. When present, the second isomer was readily removed by flash chromatography or recrystallization of the product from acetonitrile. The products in this report all contain more than one stereochemical center. The main stereochemical information on the homogeneity of a product formed, is based <sup>13</sup>C and <sup>1</sup>H NMR spectroscopy (*vide infra*). Additional information is provided by chromatography (capillary GLC).

For the hydrolytic cleavage of the bislactim ether unit in **4** to furnish the bridged amino acid esters **5**, the mild acid conditions were used; 0.25 M HCl in aqueous acetonitrile. The second product in this reaction, *viz.* valine methyl ester was removed from the crude hydrolysates by slow bulb-to-bulb distillation at 35 °C/0.05 torr.



Scheme 2

For subsequent use as reactants in solid phase peptide synthesis the amino acid esters **5** were *N*-protected and the ester groups hydrolyzed. A solution of the amino acid ester in dioxane was treated with a 50% excess 9-fluorenylmethoxycarbonyl chloride (Fmoc-Cl) and aqueous sodium bicarbonate. Fmoc-*N*-acylated amino functions are resistant towards acid hydrolysis. Consequently, HCl in dioxane was used to effect selective ester hydrolysis with formation of the *N*-protected amino acids **7** in high yields.



Scheme 3

## EXPERIMENTAL

$^1\text{H}$  NMR spectra were recorded at 300 MHz and  $^{13}\text{C}$  at 75 MHz on a Varian Unit 300. The mass spectra under electron impact conditions were recorded at 70 eV ionizing potential. The mass spectra are given as  $m/z$  (% rel. int.). IR spectra were recorded in chloroform. Analytical GLC was by a Shimadzu gas chromatograph, GC-14A (FID detector) equipped with a SPM<sup>TM</sup>-1, Silica capillar column, 30 m, 0.25 mm ID. IR spectra were recorded in chloroform.

The products were purified by flash chromatography on silica gel 60 (0.063 mm, 230–400 mesh ASTM).

**(2*R*,5*S*)-5-(2-Chloroethyl)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazine (2).** 1-Bromo-2-chloroethane (6.33 g, 44.15 mmol) in THF (30 ml) was added with stirring to a solution prepared from (2*R*,5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazine (3.5 g, 17.66 mmol), THF (200 ml), *n*-butyllithium (1.6 N, 14.14 ml, 19.43 mmol) and DMEU (4.03 g, 35.31 mmol) under argon at  $-78^\circ\text{C}$ . Stirring was continued at  $-78^\circ\text{C}$  for 18 h. The excess of 1-bromo-2-chloroethane was removed from the crude mixture by evaporation at  $40^\circ\text{C}/0.01$  torr. The product was isolated after flash chromatography on silica gel using hexane:Et<sub>2</sub>O 20:1; yield 2.44 g (53%), d.e. 90% (capillary GLC). HRMS:  $M$  260.1291. Calc. for  $\text{C}_{12}\text{H}_{21}\text{N}_2\text{O}_2\text{Cl}$  260.1284. IR:  $\nu$  1690  $\text{cm}^{-1}$  (C=N).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.68 and 1.06 (2 d,  $J$  7.7 Hz, 6H,  $-\text{CH}(\text{CH}_3)_2$ ), 1.34 (s, 3H, 2-CH<sub>3</sub>), 1.97–2.37 (m, 3H,  $-\text{CH}_2-\text{CH}_2\text{Cl}$  and  $-\text{CH}(\text{CH}_3)_2$ ), 3.21–3.38 (m, 2H,  $-\text{CH}_2-\text{CH}_2\text{Cl}$ ), 3.66 and 3.67 (2 s, 6H,  $-\text{OCH}_3$ ), 3.95 (d,  $J$  3.7 Hz 1H, H-2).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  16.91 and 19.30 ( $-\text{CH}(\text{CH}_3)_2$ ), 28.65 (2-CH<sub>3</sub>), 31.00 ( $-\text{CH}(\text{CH}_3)_2$ ), 40.51 ( $-\text{CH}_2-\text{CH}_2\text{Cl}$ ), 43.72 ( $-\text{CH}_2-\text{CH}_2\text{Cl}$ ), 52.19 and 52.32 ( $-\text{OCH}_3$ ), 57.29 (C-2), 61.08 (C-5), 162.49 and 164.31 (C=N).

**(2*R*,5*S*)-5-(2-Bromoethyl)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazine (3a).** (2*R*,5*S*)-5-(2-Chloroethyl)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-2-methylpyrazine (1.0 g, 3.83 mmol) and NaBr (0.47 g, 4.60 mmol) in DMF (30 ml) was stirred at  $70^\circ\text{C}$  for 12 h. The solvent was removed at  $40^\circ\text{C}/0.01$  torr and the product purified by flash chromatography on silica gel 60 using hexane:Et<sub>2</sub>O 20:1; yield 1.03 g (88%). HRMS:  $M$  304.0786. Calc. for  $\text{C}_{12}\text{H}_{21}\text{N}_2\text{O}_2\text{Br}$  304.0782. IR:  $\nu$  1690  $\text{cm}^{-1}$  (C=N).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.68 and 1.06 (2 d,  $J$  7.7 Hz, 6H,  $-\text{CH}(\text{CH}_3)_2$ ), 1.34 (s, 3H, 2-CH<sub>3</sub>), 1.97–2.37 (m, 3H,  $-\text{CH}_2-\text{CH}_2\text{Cl}$  and  $-\text{CH}(\text{CH}_3)_2$ ), 3.21–3.38 (m, 2H,  $-\text{CH}_2-\text{CH}_2\text{Cl}$ ), 3.66 and 3.67 (2 s, 6H,  $-\text{OCH}_3$ ), 3.95 (d,  $J$  3.7 Hz, 1H; H-2).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  16.91 and 19.30 ( $-\text{CH}(\text{CH}_3)_2$ ), 28.65 (2-CH<sub>3</sub>), 31.00 ( $-\text{CH}(\text{CH}_3)_2$ ), 40.51 ( $-\text{CH}_2-\text{CH}_2\text{Cl}$ ), 43.72 ( $-\text{CH}_2-\text{CH}_2\text{Cl}$ ), 52.19 and 52.32 ( $-\text{OCH}_3$ ), 57.29 (C-2), 61.08 (C-5), 162.49 and 164.31 (C=N).

**(2*R*,5*S*)-2,5-Dihydro-3,6-dimethoxy-5-(2-iodoethyl)-2-isopropyl-5-methylpyrazine (3b)** was prepared in 88% yield by the above procedure using sodium iodide for the halogen exchange. HRMS:  $M$  352.0649. Calc. for  $\text{C}_{12}\text{H}_{21}\text{N}_2\text{O}_2\text{I}$  352.0647. IR:  $\nu$  1690  $\text{cm}^{-1}$  (C=N).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.66 and 1.05 (2 d,  $J$  7.7 Hz, 6H,  $-\text{CH}(\text{CH}_3)_2$ ), 1.30 (s, 3H, 2-CH<sub>3</sub>), 2.12–2.49 (m, 3H,  $-\text{CH}_2-\text{CH}_2\text{I}$  and  $-\text{CH}(\text{CH}_3)_2$ ), 2.76–2.98 (m, 2H,  $-\text{CH}_2-\text{CH}_2\text{I}$ ), 3.65 and 3.66 (2 s, 6H,  $-\text{OCH}_3$ ), 3.93 (d,  $J$  4.0 Hz, 1 H, H-2).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.92 ( $-\text{CH}_2-\text{CH}_2\text{I}$ ), 16.89 and 19.30 ( $-\text{CH}(\text{CH}_3)_2$ ), 28.30 (2-CH<sub>3</sub>), 31.00 ( $-\text{CH}(\text{CH}_3)_2$ ), 45.65 ( $-\text{CH}_2-\text{CH}_2\text{I}$ ), 52.23 and 52.36 ( $-\text{OCH}_3$ ), 59.77 (C-2), 61.16 (C-5), 162.59 and 163.96 (C=N).

1,2-Bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]ethane (4a). (2*R*,5*S*)-5-Bromoethyl-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazine (2.18 g, 10.08 mmol) in THF (30 ml) was added with stirring to a solution prepared from (2*R*,5*S*/*R*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazine (2.0 g, 10.08 mmol), THF (70 ml), *n*-butyllithium (1.6 N, 6.94 ml, 11.1 mmol) and DMEU (2.30 g, 20.16 mmol) at -78 °C under argon. Stirring was continued at -78 °C for 18 h. The product was isolated after flash chromatography on silica gel 60 using hexane:Et<sub>2</sub>O 20:1; yield 2.44 g (62%), mp 96 °C (MeNO<sub>2</sub>). Capillary GLC: d.e. 96%. Found: C, 62.48; H, 9.10; N, 13.30. Calc for C<sub>22</sub>H<sub>38</sub>N<sub>4</sub>O<sub>4</sub>: C, 62.53; H, 9.06; N 13.26. HRMS: *M*: 422.2893. Calc. for C<sub>22</sub>H<sub>38</sub>N<sub>4</sub>O<sub>4</sub>: 422.2883. IR:  $\nu$  1690 cm<sup>-1</sup> (C=N). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  0.65 and 1.05 (2 d, *J* 7.7 Hz, 6 H, -CH(CH<sub>3</sub>)<sub>2</sub>), 1.28 (s, 3H, 2-CH<sub>3</sub>), 1.38–1.80 (m, 4H, -CH<sub>2</sub>-CH<sub>2</sub>-), 2.24 (dsp, *J*<sub>1</sub> 7.7 Hz, *J*<sub>2</sub> 3.7 Hz, 1 H, -CH(CH<sub>3</sub>)<sub>2</sub>), 3.63 (s, 6H, -OCH<sub>3</sub>), 3.90 (d, *J* 3.7 Hz, 1 H, H-2). <sup>13</sup>C NMR (DMSO):  $\delta$  16.72 and 19.30 (-CH(CH<sub>3</sub>)<sub>2</sub>), 24.36 (-CH<sub>2</sub>-CH<sub>2</sub>-), 28.61 (2-CH<sub>3</sub>), 30.87 (-CH(CH<sub>3</sub>)<sub>2</sub>), 52.00 and 52.02 (-OCH<sub>3</sub>), 58.31 (C-2), 61.01 (C-5), 161.62 and 165.32 (C=N).

General procedure for direct formation of bridged structures (4b, 4c). A solution of *n*-butyllithium in hexane (1.60 M solution in hexane; 3.44 ml, 5.5 mmol) was injected into a solution of (2*R*,5*S*/*R*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazine (5 mmol) and DMEU (10 mmol) in THF (50 ml). Stirring was continued for 15–20 min to complete the formation of the aza-enolate. A solution of  $\alpha,\omega$ -dibromoalkane (2.5 mmol) in THF (10 ml) was added and stirring was continued for 18 h at -78 °C. Phosphate buffer (pH 7.4) was then added, the reaction mixture allowed to warm to ambient temperature, the solvent evaporated at reduced pressure and the residual material shaken with water (30 ml) and diethyl ether (50 ml). The layers were allowed to separate, the water layer extracted twice with diethyl ether (50 ml), the combined ether extracts dried (MgSO<sub>4</sub>) and the ether distilled off. The crude product was purified by bulb-to-bulb distillation. Final purification was by flash chromatography on silica gel 60 using hexane:Et<sub>2</sub>O 20:1. The diastereomeric ratio of the crude product was determined by capillary GLC.

1,3-Bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]propane (4b) was obtained in 75% yield, mp. 65 °C (MeNO<sub>2</sub>). Capillary GLC: d.e. 88%. Found: C, 63.25; H 9.30; N, 12.78. Calc. for C<sub>23</sub>H<sub>40</sub>N<sub>4</sub>O<sub>4</sub>: C, 63.27; H, 9.23; N, 12.83. HRMS: *M*: 436.3049. Calc. for C<sub>23</sub>H<sub>40</sub>N<sub>4</sub>O<sub>4</sub>: 436.3039. IR:  $\nu$  1690 cm<sup>-1</sup> (C=N). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  0.66 and 1.05 (2 d, *J* 7.7 Hz, 6H, -CH(CH<sub>3</sub>)<sub>2</sub>), 0.71–0.82 (m, 2H, -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-), 1.27 (s, 3H, 2-CH<sub>3</sub>), 1.32–1.81 (m, 2H, -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-), 2.20 (dsp, *J*<sub>1</sub> 7.7 Hz, *J*<sub>2</sub> 4.0 Hz, 1 H, -CH(CH<sub>3</sub>)<sub>2</sub>), 3.61 (s, 6H, -OCH<sub>3</sub>), 3.87 (d, *J* 4.0 Hz, 1 H, H-2). <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  16.87 and 19.28 (-CH(CH<sub>3</sub>)<sub>2</sub>), 19.09 (-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-), 28.58 (2-CH<sub>3</sub>), 30.97 (-CH(CH<sub>3</sub>)<sub>2</sub>), 41.31 (-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-), 51.97 and 51.93 (-OCH<sub>3</sub>), 58.31 (C-2), 61.01 (C-5), 161.71 and 165.26 (C=N).

1,4-Bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]butane (4c) was obtained in 82% yield, mp. 57 °C (MeNO<sub>2</sub>). Capillary GLC: d.e. 88 %. Found: C, 63.94; H 9.42; N, 12.40. Calc. for C<sub>24</sub>H<sub>42</sub>N<sub>4</sub>O<sub>4</sub>: C, 63.97; H, 9.39; N, 12.43. HRMS: *M* 450.3206. Calc. for C<sub>24</sub>H<sub>42</sub>N<sub>4</sub>O<sub>4</sub>: 450.3196. IR:  $\nu$  1690 cm<sup>-1</sup> (C=N). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  0.66 and 1.04 (2 d, *J* 6.9 Hz, 6H; -CH(CH<sub>3</sub>)<sub>2</sub>), 0.89–1.07 (m, 4H, -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-), 1.29 (s, 3H, 2-CH<sub>3</sub>), 1.38–1.90 (m, 4H; -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-), 2.25 (dsp, *J*<sub>1</sub> 6.9 Hz, *J*<sub>2</sub> 3.4 Hz, 1H, -CH(CH<sub>3</sub>)<sub>2</sub>), 3.64 (s, 6H, -OCH<sub>3</sub>), 3.91 (d, *J* 3.4 Hz, 1 H, H-2). <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$

16.81 and 19.38 ( $-\text{CH}(\text{CH}_3)_2$ ), 24.44 ( $-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-$ ), 28.67 ( $2-\text{CH}_3$ ), 30.95 ( $-\text{CH}(\text{CH}_3)_2$ ), 41.39 ( $-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-$ ), 52.11 ( $-\text{OCH}_3$ ), 58.40 ( $\text{C}-2$ ), 61.10 ( $\text{C}-5$ ), 161.71 and 165.41 ( $\text{C}=\text{N}$ ).

General procedure for the preparation of (*S,S*)- $\alpha,\omega$ -diamino- $\alpha,\omega$ -dimethylalkanedioic acid dimethyl esters (**5**), and subsequently (*S,S*)- $\alpha,\omega$ -bis(9-fluorenylmethyloxycarbonylamino)- $\alpha,\omega$ -dimethylalkanedioic acid dimethyl esters (**6**). The  $\alpha,\omega$ -bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]alkane (**4**) (2.00 mmol) was dissolved in acetonitrile (24 ml) and 0.5 M HCl (24.0 ml) was added dropwise. The solution was stirred at ambient temperature for 12 h. Aqueous ammonia was subsequently added until pH 11, the mixture extracted (3x) with chloroform (20 ml), the chloroform solution dried ( $\text{MgSO}_4$ ), evaporated and the valine methyl ester removed from the residual product by slow bulb-to-bulb distillation at 35 °C/0.05 torr. 9-Fluorenylmethyloxycarbonyl chloride (3.66 mmol) and 1 M  $\text{NaHCO}_3$  (3.7 ml) were added gradually together with stirring to a solution of the crude (*S,S*)- $\alpha,\omega$ -diamino- $\alpha,\omega$ -dimethylalkanedioic acid dimethyl ester (1.22 mmol) in dioxane (30 ml). The mixture was stirred at ambient temperature for 20 h before the dioxane was removed by distillation. The residue was extracted (3x) with chloroform (20 ml), the chloroform solution washed, dried ( $\text{MgSO}_4$ ), the solvent distilled off and the product isolated from the residual material after flash chromatography using hexane:AcOMe from 4:1 to 2:1.

(*S,S*)-2,5-Bis(9-fluoromethyloxycarbonylamino)-2,5-dimethylhexanedioic acid dimethyl ester (**6a**). Acid hydrolysis of 1,2-bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]ethane gave the intermediate amino acid ester (*S,S*)-2,5-diamino-2,5-dimethylhexanedioic acid dimethyl ester (**5a**) in 91% yield as a non-solid material. IR:  $\nu$  1731  $\text{cm}^{-1}$  ( $\text{C}=\text{O}$ ).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.08–1.24 (m, 1H,  $\text{CH}_2$ ), 1.30 (s, 3H,  $-\text{CH}_3$ ), 1.42–1.76 (m, 1H,  $\text{CH}_2$ ,  $-\text{NH}_2$ ), 3.69 (s, 3H,  $-\text{OCH}_3$ ).

The crude amino acid ester **5a** was acylated to yield the title compound **6a** in 75% yield as a non-solid. Found: C, 71.02; H, 6.26. Calc. for  $\text{C}_{40}\text{H}_{40}\text{N}_2\text{O}_8$ : C, 70.99; H, 5.96. IR:  $\nu$  1738, 1719  $\text{cm}^{-1}$  ( $\text{C}=\text{O}$ ).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.56 (s, 3H,  $-\text{CH}_3$ ), 1.03, 1.24, 1.73 and 2.11 (br s, 4 H,  $-\text{CH}_2-\text{CH}_2-$ ), 3.74 (s, 3H,  $-\text{OCH}_3$ ), 4.22 (t,  $J$  6.2 Hz, 1H,  $-\text{Fmoc}$ ), 4.37 (br s, 2H,  $-\text{CH}_2-\text{Fmoc}$ ), 5.59 (br s, 1H,  $-\text{NH}-\text{Fmoc}$ ), 7.23–7.81 (m, 8H,  $\text{Fmoc}$ ).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  23.36, 23.90, 36.57, 47.21, 52.68, 59.88, 66.32, 119.90, 124.94, 126.97, 127.90, 141.27, 143.87, 174.66.

(*S,S*)-2,6-Bis(9-fluoromethyloxycarbonylamino)-2,6-dimethylheptanedioic acid dimethyl ester (**6b**). Acid hydrolysis of 1,3-bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]propane gave the intermediate amino acid ester (*S,S*)-2,6-diamino-2,6-dimethylheptanedioic acid dimethyl ester (**5b**) in 91% as a non-solid material. IR:  $\nu$  1731  $\text{cm}^{-1}$  ( $\text{C}=\text{O}$ ).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.02–1.34 (m, 2H,  $\text{CH}_2$ ), 1.30 (s, 3H,  $-\text{CH}_3$ ), 1.34–1.82 (m, 4H,  $\text{NH}_2$ ,  $\text{CH}_2$ ), 3.69 (s, 3H,  $\text{OCH}_3$ ).

The crude amino acid ester **5b** was acylated to yield the title compound **6b** in 75% yield as a non-solid. IR:  $\nu$  1737, 1706  $\text{cm}^{-1}$  ( $\text{C}=\text{O}$ ).  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  1.56 (s, 3H,  $-\text{CH}_3$ ), 1.03, 1.24, 1.73 and 2.11 (br s, 6H,  $-\text{CH}_2-\text{CH}_2-\text{CH}_2-$ ), 3.74 (s, 3H,  $-\text{OCH}_3$ ), 4.22 (t,  $J$  6.2 Hz, 1H,  $-\text{Fmoc}$ ), 4.37 (br s, 2H,  $-\text{CH}_2-\text{Fmoc}$ ), 5.59 (s, 1H,  $-\text{NH}$ ), 7.23–7.81 (m, 8H,  $\text{Fmoc}$ ).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  23.36, 36.31, 47.16, 52.66, 59.66, 66.43, 119.93, 124.94, 127.02, 127.63, 141.26, 143.85, 171.51.

(S,S)-2,7-Bis(9-fluoromethyloxycarbonylamino)-2,7-dimethyloctanedioic acid dimethyl ester (6c). Acid hydrolysis of 1,4-bis[(2*R*, 5*S*)-2,5-dihydro-3,6-dimethoxy-2-isopropyl-5-methylpyrazin-5-yl]butane gave the intermediate amino acid ester (S,S)-2,7-Diamino-2,7-dimethyloctanedioic acid dimethyl ester (5c) in 92% yield as a non-solid material. IR:  $\nu$  1730  $\text{cm}^{-1}$  (C=O).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.02–1.28 (m, 2H,  $\text{CH}_2$ ), 1.29 (s, 3H,  $-\text{CH}_3$ ) 1.42–1.85 (m, 4H,  $-\text{NH}_2$ ,  $\text{CH}_2$ ), 3.69 (s, 3H,  $-\text{OCH}_3$ ).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  24.36 ( $-\text{CH}_2\text{CH}_2-$ ), 26.27 ( $-\text{CH}_3$ ), 40.90 ( $-\text{CH}_2\text{CH}_2-$ ), 52.05 ( $-\text{OCH}_3$ ), 57.65 ( $-\text{CNH}_2$ ), 178 ( $-\text{C}=\text{O}$ ).

The crude amino acid ester **5c** was acylated to yield the title compound **6c** in 88% yield as a non-solid. IR:  $\nu$  1738, 1707  $\text{cm}^{-1}$  (C=O).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.59 (s, 3H,  $-\text{CH}_3$ ), 1.02, 1.55 1.75 and 2.12 (br s, 4H;  $-\text{CH}_2\text{CH}_2-$ ), 3.74 (s, 3H,  $-\text{OCH}_3$ ), 4.21 (t,  $J$  6.2 Hz, 1H,  $-\text{Fmoc}$ ), 4.37 (br s, 2H,  $-\text{CH}_2\text{Fmoc}$ ), 5.59 (br s, 1H,  $-\text{NH}$ ), 7.21–7.79 (m, 8H,  $\text{Fmoc}$ ).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  23.36, 23.90, 36.56, 47.20, 52.68, 59.87, 66.32, 119.90, 124.93, 126.96, 127.59, 141.26, 143.86, 174.65.

General procedure for the preparation of (S,S)- $\alpha,\omega$ -bis(9-fluorenylmethyloxycarbonylamino)- $\alpha,\omega$ -dimethylalkanedioic acids (7). A solution of (S,S)- $\alpha,\omega$ -bis(9-fluorenylmethyloxycarbonylamino)- $\alpha,\omega$ -dimethylalkanedioic acid dimethyl ester (1.00 mmol) in dioxane (10 ml) and 6 M HCl (0.67 ml, 4.00 mmol) was heated at 90  $^\circ\text{C}$  for 24 h. The dioxane was then distilled off at reduced pressure, the residue extracted with chloroform (2x20 ml), the solution dried ( $\text{MgSO}_4$ ), evaporated and the residual material subjected to flash chromatography using hexane:AcOEt:AcOH (1:1:0.1). Fractions containing the product were pooled, concentrated to a small volume at reduced pressure and dried overnight to give the title compound as a white powder in high vacuum.

(S,S)-2,5-Bis(9-fluorenylmethyloxycarbonylamino)-2,5-dimethylhexanedioic acid (7a) was obtained from (S,S)-2,5-bis(9-fluorenylmethyloxycarbonylamino)-2,5-dimethylhexanedioic acid dimethyl ester in 71% yield of a non-crystalline solid. Found: C, 70.47; H, 5.58. Calc. for  $\text{C}_{38}\text{H}_{36}\text{N}_2\text{O}_8$ : C, 70.36; H, 5.59.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  1.13 (s (br), 1 H,  $-\text{CH}_2\text{CH}_2\text{CH}_2-$ ), 1.30 (s, 3H,  $-\text{CH}_3$ ), 1.68 (s (br), 1H,  $-\text{CH}_2\text{CH}_2-$ ), 4.22 (m, 3H,  $-\text{Fmoc}$ ), 7.21–7.96 (m, 9H,  $-\text{Fmoc}$  and  $-\text{NH}$ ), 12.40 (s (br), 1H,  $-\text{COOH}$ ).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  22.97, 23.98, 37.09, 44.00, 47.17, 58.76, 65.59, 120.46, 125.63, 127.44, 127.98, 141.10, 144.27, 155.01, 175.71.

(S,S)-2,6-Bis(9-fluorenylmethyloxycarbonylamino)-2,6-dimethylheptanedioic acid (7b) was obtained from (S,S)-2,6-bis(9-fluorenylmethyloxycarbonylamino)-2,6-dimethylheptanedioic acid dimethyl ester in 88% yield of a non-crystalline solid. Found: C, 70.56; H, 5.80. Calc. for  $\text{C}_{39}\text{H}_{38}\text{N}_2\text{O}_8$ : C, 70.68; H, 5.78.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  1.16 (s (br), 2 H,  $-\text{CH}_2\text{CH}_2-$ ), 1.30 (s, 3H,  $-\text{CH}_3$ ), 1.68 (s (br), 2 H ( $-\text{CH}_2\text{CH}_2\text{CH}_2-$ ), 4.22 (m, 3H,  $-\text{Fmoc}$ ), 7.21–7.98 (m, 9 H,  $-\text{Fmoc}$  and  $-\text{NH}$ ), 12.14 (s (br), 1H,  $-\text{COOH}$ ).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  21.47, 22.95, 23.98, 37.10, 47.16, 58.75, 65.60, 120.46, 125.65, 127.45, 128.00, 141.10, 144.27, 155.06, 172.99, 175.70.

(S,S)-2,7-Bis(9-fluorenylmethyloxycarbonylamino)-2,7-dimethyloctanedioic acid (7c) was obtained from (S,S)-2,7-bis(9-fluorenylmethyloxycarbonylamino)-2,7-dimethyloctanedioic acid dimethyl ester in 88% yield of a non-crystalline solid. Found: C, 70.97; H, 6.07. Calc. for  $\text{C}_{40}\text{H}_{40}\text{N}_2\text{O}_8$ : C, 70.99; H, 5.96.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  1.16 (s (br), 2H,  $-\text{CH}_2\text{CH}_2-$ ), 1.30 (s, 3H,  $-\text{CH}_3$ ), 1.68 (s (br); 2H,  $-\text{CH}_2\text{CH}_2\text{CH}_2-$ ), 4.22

(m, 3H, -Fmoc), 7.21 - 7.98 (m, 9 H, -Fmoc and -NH), 12.14 (s (br), 1H, -COOH).  $^{13}\text{C}$  NMR (DMSO- $d_6$ ):  $\delta$  21.47, 22.95, 23.98, 37.10, 47.16, 58.75, 65.60, 120.46, 125.65, 127.45, 128.00, 141.10, 144.27, 155.06, 172.99, 175.70.

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